



UNIVERSITI PUTRA MALAYSIA

**SYNTHESIS AND CHARACTERISATION OF VANADIUM
PHOSPHOROUS OXIDE CATALYSTS**

LOOI MING HOONG.

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By

LOOI MING HOONG

**Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia in
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February 2004



Specially dedicated to

*Husband, McCool, Anderson, Chawla, Clark, Ramon and Brown
the seven brave souls who perished on board the Columbia.*

Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfilment of the requirements for Doctor of Philosophy

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February 2004

Chairman: Taufiq Yap Yun Hin, Ph.D.

Faculty: Science and Environmental Studies

The V/P/O catalysts were prepared by using (a) organic method (VPO); (b) via dihydrate phase (VPD) and (c) aqueous method (VPA). The effect of calcination time, preparation method and addition of metal cations as dopants to the physico-chemical properties were studied using nitrogen physisorption measurements, scanning electron microscopy (SEM), X-ray diffraction, inductively coupled plasma (ICP) spectroscopy, redox titration, temperature programmed desorption (TPD) and temperature programmed reduction (TPR).

Surface areas of these mesoporous vanadium phosphorous oxides were apparently influenced by the length of calcination time, preparation method and incorporation of metal cations. The changes of surface areas were related to the changes of surface and bulk morphologies as evidenced by SEM.

X-ray diffraction revealed that while all of the V/P/O were consisted of predominantly V^{4+} phase of $(VO)_2P_2O_7$, minority V^{5+} phase which resumed many crystalline forms made up the remaining portion. Addition of metal cations to the basic matrix resulted

in slight loss of crystallinity for V/P/O by VPO method but a huge effect to V/P/O by VPD method which also accompanied by a marked increase in lattice strains.

Average oxidation state of vanadium (determined by redox titration) can be altered by (i) increase time of calcinations, (ii) predisposed by the preparation method and (iii) incorporation of metal cations.

The persistently lowering of the amount of oxygen atoms that were available thermally until stabilisation at around half a monolayer suggests that the catalytic activity, and hence the conversion of hydrocarbon is stabilised after 100 h. From the extra peak that appeared at higher temperature during TPR by H_2 for V/P/O calcined for longer duration, it was postulated that the selective nature of equilibrated V/P/O is originated from these oxygen atoms.

Based on the above argument, it was suggested that the VPD type of preparation would result in V/P/O with more active and selective nature than VPO and VPA. The addition of metal cations to bulk VPO was shown to increase the activity in the order of $Zn > VPO (bulk) > Cr > Co$ and selectivity in the order of $Co > Cr > Zn > VPO (bulk)$. While for VPD the activity and selectivity will be in the order of $Co > VPD (bulk) > Cr > Zn$ and $Zn > Cr > VPD (bulk) \approx Co$, respectively.

Abstrak thesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk Doktor Falsafah

SINTESIS DAN PENCIRIAN MANGKIN VANADIUM FOSFORUS OKSIDA

Oleh

LOOI MING HOONG

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Mangkkin V/P/O telah disintesis dengan menggunakan (a) kaedah organik (VPO); (b) melalui fasa dihidrat (VPD) dan (c) kaedah akuas (VPA). Kesan daripada tempoh pengkalsinan, kaedah sintesis dan penambahan kation-kation logam sebagai penggalak kepada ciri-ciri fiziko-kimia telah dikaji dengan menggunakan kaedah fizikjerapan nitrogen, mikroskop imbasan elektron (SEM), pembelauan sinar-X, spektroskopi plasma pasangan induktif (ICP), penitratan redoks, penyahjerapan suhu berprogram (TPD) dan penurunan suhu berprogram (TPR).

Luas permukaan oksida-oksida vanadium fosforus yang berciri liang meso ini dipengaruhi oleh tempoh pengkalsinan, cara sintesis dan juga penambahan kation-kation logam. Perubahan luas permukaan ini boleh dikaitkan dengan perubahan morfologi permukaan dan pukal seperti yang dikesan oleh SEM.

Pembelauan sinar-X menunjukkan bahawa semua V/P/O terdiri daripada fasa V^{4+} dalam bentuk $(VO)_2P_2O_7$ sebagai fasa majority dan fasa minoriti V^{5+} yang mempunyai pelbagai bentuk kristal. Sel unit untuk prekursor dan mangkkin telah ditentukan sebagai ortorombik dengan pemalar-pemalar kekisi yang menyerupai bahan piawai.

Penambahan kation-kation logam kepada matriks asas menyebabkan sedikit pengurangan dalam kekristalan untuk V/P/O yang disintesis melalui kaedah VPO. Kesan yang nyata kepada sifat kekristalan dan penambahan regangan kekisi V/P/O dengan kaedah VPD telah diperhatikan.

Keadaan pengoksidaan purata untuk vanadium (dengan penitratan redoks) boleh diubah dengan (i) menambahkan tempoh pengkalsinan, (ii) kaedah sintesis dan (iii) penambahan kation-kation logam.

Atom-atom oksigen yang diperolehi secara terma sentiasa berkurangan sehingga tahap stabil pada lebih kurang setengah mono-lapisan mencadangkan bahawa aktiviti mangkin dan seterusnya penukaran hidrokarbon akan stabil selepas 100 j. Daripada puncak tambahan yang muncul pada suhu yang lebih tinggi semasa TPR dengan hydrogen untuk V/P/O yang dikalsinkan dalam masa yang lebih panjang, cadangan bahawa sifat pemilihan V/P/O dalam keadaan keseimbangan berasal daripada atom-atom oksigen ini telah dibuat.

Berasaskan kepada perbincangan di atas, kaedah VPD akan menghasilkan V/P/O yang bersifat lebih aktif dan memilih daripada kaedah VPO dan VPA. Penambahan kation-kation logam kepada pukal VPO akan meningkatkan aktiviti dalam susunan $\text{Zn} > \text{VPO (pukal)} > \text{Cr} > \text{Co}$ dan pemilihan dalam susunan $\text{Co} > \text{Cr} > \text{Zn} > \text{VPO (pukal)}$. Semetara untuk VPD pula, aktiviti dan pemilihan akan berada dalam susunan $\text{Co} > \text{VPD (pukal)} > \text{Cr} > \text{Zn}$ dan $\text{Zn} > \text{Cr} > \text{VPD (pukal)} \approx \text{Co}$ masing-masing.

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I certify that an Examination Committee met on 12th February 2004 to conduct the final examination of Looi Ming Hooi on his Doctor of Philosophy thesis entitled "Synthesis and Characterisation of Vanadium Phosphorous Oxide Catalysts" in accordance with Universiti Pertanian Malaysia (Higher Degree) Act 1980 and Universiti Pertanian Malaysia (Higher Degree) Regulations 1981. The Committee recommends that the candidate be awarded the relevant degree. Members of the Examination Committee are as follows:

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DECLARATION FORM

I hereby declare that the thesis is based on my original work except for quotations and citations which have been duly acknowledged. I also declare that it has not been previously or concurrently submitted for any other degree at UPM or other institutions.



LOO MING HOONG

Date: 7 JUNE 2004

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